# 1,3-Dipolar Cycloadditions of Diazoalkanes to Pyridazines. Asymmetric 1,3-Dipolar Cycloadditon of Azomethine Imines Derived from Diazoalkane-Pyridazine Cycloadducts

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# i) Introduction.

Since the discovery of aliphatic diazo compounds towards the end of the previous century [1-4], they have played an important role in 1,3-dipolar cycloaddition reactions [5] to the systems with double and triple bonds, heteromultiple bonds and heterocumulenes to form five-membered rings [6]. In aromatic and heteroaromatic series, there are only sporadic examples of cycloadditions to five- and six-membered rings reported in the literature, since these systems are less reactive and therefore the diazoalkanes have been frequently used as a source for carbenes in cyclopropanation reactions to produce norcaradienes and cycloheptatrienes and their aza-analogs [6].

In heterocycles a regiospecific cycloadditions to some highly substituted nitrogen-containing systems, such as imidazoles [7], pyridinones [8], diazepines [9] and oxygen-containing heterocycles, such as furanones [10-13] have been reported. In azolo[1,5-a]pyridine series the formation of bis-cycloadducts has been observed [14] and in the pyrrolo[1,2-a]pyrimidine series three cycloadducts have been isolated [15].

# ii) Cycloadditions of Diazoalkanes to Pyridazines.

Cycloaddition of diazomethane to monocyclic pyridazine derivatives and bicyclic azolo- and azinopyridazines with a bridgehead nitrogen atom, of general formula 1

produces CH,CH-dihydro cycloadducts 2, which can be transformed by sigmatropic rearrangements into CH,NHand NH,NH-dihydro cycloadducts 3-5. The reaction proceeds regiospecifically to the partially localized and polarized C<sub>4</sub>-C<sub>5</sub> double bond in monocyclic pyridazines or to  $C_7$ - $C_8$  double bond in azolopyridazines [16,17]. With less reactive diazoalkanes, such as 2-diazobutane, phenyldiazomethane, 1-diazo-1-phenylethane the corresponding primary CH,CH-dihydro products of the type 2, and the rearranged NH,CH-dihydro cycloadducts of the types 3 and 4 have been isolated [18]. By cycloaddition of diazomethane the formation of the primary CH,CH-dihydro cycloadduct is followed by dehydrogenation and sigmatropic rearrangement giving the tautomeric intermediates 6 and 7. N-Methylation of these tautomers with an excess of diazomethane yields mixtures of the corresponding isomeric pairs 8 ( $R_1 = Me$ ) and 9 ( $R_1 = Me$ ) [19-21] (Scheme 1).

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Due to this additional tautomerization and further methylation, systematic studies of 1,3-dipolar cycloadditions to pyridazines and fused pyridazines have been carried out with 2-diazopropane in order to avoid further transformations.

It has been reported that regiospecific 1,3-dipolar cycloaddition of 2-diazopropane to 2-methyl-6-phenylpyridazin-3(2H)-one (10) in diethyl ether gives a mixture of three products: 1,2-diazepine derivative 13, 4-isopropyl derivative 14 and diazabicyclo[4.1.0]heptanone derivative 15, presumably formed from the primary cycloadduct 11 [22]. When this reaction was reinvestigated, it turned out, that it is strongly dependent upon the solvent and temperature. The cycloadduct 11 could be isolated in pure form at temperatures below 0°, due to its low solubility in diethyl ether. At room temperature, the secondary reactions were observed, in which the intermediate 11 is transformed in three different ways. In the presence of an acid, the isomerization into NH,NH-dihydro intermediate 12 takes place. In the presence of oxygen from the air dehydrogenation is the main process producing pyrazolo[3,4-d]pyridazine derivative 16, while elimination of a molecule of nitrogen from the pyrazole part of the bicyclic system followed by rearrangement, gives a mixture of 13-15. The relative proportion of the products formed according to these three pathways is strongly dependent upon the solvent used in the reaction. In polar solvents, such as dimethylformamide, methanol and ethanol, in the presence of air, 11 is quantitatively dehydrogenated affording 16, while in less polar solvents, in which oxygen is less soluble, such as acetonitrile, dioxane, ethyl acetate, benzene, and other compounds, 13-15 are formed [23] (Scheme 2).

Oxidation of 11 and 12 could be achieved with a variety of oxidizing agents, such as oxygen and bromine. The simplest method is oxidation by air in the presence of a base. The transformation is practically quantitative to give 16 as the only product. Catalytic hydrogenation of 16 over Pd/C produces 11 as the only product [23].

# iii) Directed Regiospecificity.

The cycloaddition of diazomethane to pyridazine derivatives has been first observed as a side reaction by methylation of 6-hydroxypyridazin-3(2H)-one 17 in which besides O-methyl- 19 and (O,N)-dimethyl- 20 also the corresponding pyrazolo[3,4-d]pyridazine derivative 18 was formed in 9% yield [24,25] (Scheme 3). N-Methylpyridazin-3(2H)-ones 21 give with diazomethane N-methylpyrazolo[3,4-d]pyridazin-4(5H)-ones 22 and 23 as the major products and isomeric -7(6H)-one derivatives 24, as the minor product [26] (Scheme 3).

1,3-Dipolar cycloadditions of diazoakanes 4- and 5-unsubstituted pyridazin-3(2H)-ones is regiospecific producing in most cases pyrazolo[3,4-d]pyridazin-4(5H)-ones as the major products and in some instances the corresponding -7(6H)-ones as the minor products. However, the cycloadditions are regiospecifically controlled by the position of substituents. When 4-substituted 6-methoxy-2-methylpyridazin-3(2H)-ones 26 are treated with 2-diazopropane in a mixture of chloroform and diethyl ether in the presence of triethylamine the corresponding 3H-pyrazolo[3.4-d]pyridazin-7(6H)-one 28 is formed as the only product after elimination of a molecule of HX from the primary cycloadducts 27. On the other hand, 5-substituted derivatives 29 afford the isomeric -4(5H)-ones 31 as the only product, while 6-methoxy-2-methylpyridazin-3(2H)-one (25) gives a mixture of 28 (5%) and 31 (82%) [27,28] (Scheme 4).

Scheme 2

# Scheme 4

X = Cl, SPh, SOPh

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Cycloadditions of diazoalkanes to bicyclic heteroaromatic  $10\pi$ -electron systems with a bridgehead nitrogen atom, such as imidazo[1,2-b]pyridazine, s-triazolo[4,3-b]pyridazine, s-triazolo[1,5-b]pyridazine, and tetrazolo[1,5-b]pyridazine proceed as a regiospecific cycloaddition to partially localized and polarized double bond C7-C8 followed by loss of a molecule of hydrogen from the primary or rearranged cycloadducts affording stable pyrazoloazolopyridazines. In this manner derivatives of the following systems were prepared: 9H-imidazo[1,2-b]pyrazolo[4,3-d]pyridazine [29-31], 9H-pyrazolo[4,3-d]-s-triazolo[4,3-b]pyridazine [32], 9H-pyrazolo[4,3-d]-s-triazolo[1,5-b]pyridazine [32] and 9H-pyrazolo[4,3-d]tetrazolo[1,5-b]pyridazine [33]. Similarly, cycloaddition takes place to the  $C_8$ - $C_9$  double bond in pyrimido[1,2-b]pyridazin-4(10H)-one 32 giving 10H-pyrazolo[4,3-d]pyrimido-[1,2-b]pyridazin-4(10H)-one derivatives 33 [34] (Scheme 5).

Scheme 5

Scheme 5

$$R_3$$
 $R_2$ 
 $R_3$ 
 $R_4$ 
 $R_3$ 
 $R_4$ 
 $R_5$ 
 $R_5$ 

iv) The Synthesis of Isomeric Pyrazolo[3,4-d]azolopyridazines.

Derivatives of pyrazolo[3,4-d]azolopyridazine fused systems can be prepared by azido-tetrazolo valence isomerization, observed earlier in tetrazolo[1,5-b]pyridazine system [35-39], from the corresponding pyrazolo [4,3-d]tetrazolo[1,5-b]pyridazine systems. For example, the compound 34 gives with 2-diazopropane the corresponding 6-chloropyrazolo[4,3-d]tetrazolo[1,5-b]pyridazine 35. This was transformed by heating with hydrazine into hydrazino compound 36 and further transformed with nitrous acid into azido derivative 37. When this is heated in dimethyl sulfoxide solution at 110°, the azido-tetrazolo isomerization produces a mixture of 37 and isomeric 7H-pyrazolo[3,4-d]tetrazolo[1,5-b]pyridazine derivative 38 in a ratio 4:1. In the reaction of 36 with a mixture of triethyl orthoformate and acetic anhydride 7H-pyrazolo[3,4-d]-striazolo[4,3-b]pyridazine derivative 39 was obtained. Similarly, derivatives of imidazo[1,2-b]pyrazolo[3,4-d]pyridazine 41 were prepared [20,21] (Scheme 6).

Cycloaddition of 2-diazopropane to tricyclic systems, such as 42 occurs across  $C_4$ - $C_5$  double bond producing a mixture of the isomeric derivatives of 43 and 44 of the tetracyclic system 11*H*-pyrazolo[3,4-*d*]-bis-s-triazolo-[4,3-b:3',4'-f]pyridazine. The structure of both isomers were proven by independent syntheses starting from 45 and 48 by the reaction sequences  $45 \rightarrow 46 \rightarrow 47 \rightarrow 43$  and  $48 \rightarrow 49 \rightarrow 50 \rightarrow 51 \rightarrow 44$ , respectively [40]. Similarly, the tricyclic system 52 gives derivatives of two isomeric tetracyclic systems 53 and 54 [33] (Scheme 7).

- v) Transformations of Cycloadducts.
- a) Thermal Reactions.

3,3-Dimethyl-3H-pyrazolo[3,4-d]pyridazin-4(5H)-ones 55, when heated in polyphosphoric acid at 120° for 30 minutes, give the isomeric  $N_2$  methylated products 56 and  $C_{3a}$ -methylated isomers 57. The isomeric -7(6H)-one 58 affords only the  $N_2$ -methylated product 59 [41]. From the NH,NH-dihydro derivative 80 elimination of methane takes place by heating in an inert solvent giving 1H-pyrazolo[3,4-d]pyridazine derivative 61 [23] (Scheme 8).

# b) Photochemical Reactions.

The irradiation of compounds 62 leads to the loss of molecular nitrogen from the pyrazole part of the molecule in the first step to produce the diradical 63. In the second step, when the reactions were carried out in a mixture of tetrahydrofuran and pentane 8-isopropenylazolopyridazines 65 are formed in 7-29% yield. Irradiation in methanol provides mixtures of products, from which 65 are isolated as the minor components (5-30%) and methyl ethers 66 as the major components (36-67%). Cyclopropa[d]pyridazine derivatives 64 were not isolated (Scheme 9). However, the 1,3-diradicals 63 react also with furan to form 1:1 adducts 67 in 56-70% yields, accompanied by small amounts of the alkenes 65 (8-12%). In the presence of buta-1,3-diene two types of addition products are formed. The major components are 7H-dihydrocyclohepta[d]pyridazine 68, corresponding to 1,4-cycloaddition. The minor components correspond to 1,2-cycloaddition to form cyclopenta[d]pyridazines 69 [32,34,42,-43,44] (Scheme 9).

# c) Ring Enlargements.

The cycloaddition of 2-diazopropane to 6-methoxy-2-methyl-5-phenylsulfonylpyridazin-3(2H)-one (70) produces the intermediate 71, from which phenylsulfinic acid is not eliminated, contrary to the observation with other 5-substituted pyridazin-3(2H)-one derivatives. In this case, elimination of molecular nitrogen followed by rearrangement affords 1,2-diazepine derivative 72. This reacts further, when an excess of 2-diazopropane is used, to give pyrazolo[3,4-d]-1,2-diazepine derivative 73 [27] (Scheme 10).

Scheme 6

MeO

MeO

# vi) Tranformations of Dihydrocycloadducts.

# a) Formation of Azomethine Imines.

Azomethine imines are important intermediates as 1,3-dipoles in cycloadditions and electrocyclic reactions in which five-, six-, and seven-membered rings are formed [45].

The dihydro intermediates 74 and 75 react with aldehydes and masked aldehydes, such as N,N-dimethylformamide dimethyl acetal, and dimethyl acetylenedicarboxylate to give azomethine imines 76 and 77 [23] (Scheme 11).

# b) Thermal Rearrangements of Azomethine Imines.

An interesting thermal rearrangement was observed, when studying the chemical properties of these intermediates. Azomethine imines 78, prepared from dihydro cycloadducts and ketones, are transformed by heating in xylene, into tetrazolopyridazinodiazepine derivatives 79 [46]. Similarly, the azomethine imine 80, prepared from dihydro cycloadduct and cyclohexanone, gives the spiro compound 81 [47] (Scheme 12).

# Scheme 12 **79 78** $R_3$ $R_2$ $R_3$ $R_2$ Me Me Me Me Me Me Me CD<sub>3</sub> CD<sub>3</sub> H $CD_3$ Me Me CD<sub>3</sub> Me Me Me Me Et Me 81 80

# c) 1,3-Dipolar Cycloadditions of Azomethine Imines.

Azomethine imines react as 1,3-dipoles with unsaturated compounds, such as olefins, acetylenes and arynes. The azomethine imines can be prepared *in situ*. For example, compound **82** reacts with a mixture of formaldehyde and an unsaturated compound to give tetracyclic systems **83-87**, and with a mixture of benzaldehyde and benzyne to form the pentacyclic system **88** [47] (Scheme 13).

This reaction sequence can be conveniently applied to the synthesis of γ-amino acid derivatives. For example, the NH,NH-dihydro cycloadduct 82 forms with the protected aminoacetaldehyde the corresponding azomethine imine 89, which reacts with dimethyl acetylenedicarboxylate or dimethyl maleinate to give the corresponding pyrazolo[1',2':1,2]pyrazolo[4,3-d]tetrazolo[1,5-b]pyridazine derivatives 90 and 91 [47], respectively. The structure of 91 was also confirmed by X-ray analysis [48] (Scheme 14).

# d) Asymmetric 1,3-Dipolar Cycloadditions of Azomethine Imines.

The development of asymmetric 1,3-dipolar cycloaddition reactions has in recent years entered a new stage. The selectivity challenge is to control regiodiastereo-, and enantioselectivity of this type of reactions.

There are four types of carbohydrate derivatives, which reacts as 1,3-dipoles in cycloaddition reactions, described in the literature. They are azides [49,50], nitrones [51-53], nitrile oxides [54], and diazo derivatives [55,56]. Only azides and acyclic diazo carbohydrates have been isolated in pure form, while the corresponding nitrones and nitrile oxides have been prepared *in situ* and used as such in further transformations.

The use of azomethine imines in asymmetric 1,3-dipolar cycloadditions with alkenes is limited [57].

A new type of stable azomethine imines derived from 6-chloro-7,8-dihydro-9,9-dimethyl-9H-pyrazolo[4,3-b]tetrazolo[1,5-b]pyridazine (82) and carbohydrate derived aldehydes, such as tetra- and penta-O'-substituted aldehydo sugars: penta-benzoyl-al-D-glucose, tetraacetyl-al-L-arabinose and pentaacetyl-al-D-galactose to give the chiral azomethine imines 92, 94, and 97, respectively. These, when heated with methyl acrylate as dipolarophile in acetonitrile produced the corresponding O'-acylated compounds 93, 95, and 98, respectively. Compounds 95 and 98 were formed as pure (>95% d.e.) stereoisomers, while 93 contains around 10% of another isomer. The reaction can be carried out also as a one-pot synthesis. In this manner, the compound 82 was treated with D-ribose and methyl acrylate by heating in methanol in the presence of catalytic amounts of trifluoroacetic acid to give 96 [58,59] (Scheme 15). On the basis of this latter observation the reaction was extended to some other protected and unprotected sugars to give compounds 100-105 as pure (>95% d.e.) stereoisomers [60] (Scheme 16).

In all these reactions two new chiral centers at  $C_8$  and  $C_{10}$  are formed. Since the absolute configuration at  $C_{1'}$  in the final product is given by the absolute configuration at  $C_2$ , *i.e.* at C atom  $\alpha$  to the aldehydo group of the starting carbohydrate, the absolute configurations at  $C_8$  and  $C_{10}$  can be determined by  $^1H$  nmr data on the basis of the magnitude of coupling constants. It turned out, that compounds 106, derived from  $^2R$ -carbohydrates, *i.e.* carbohydrates with R-configuration at the C atom next to the aldehydo group, have the absolute configuration (1'S,8S,10S), while the compounds 107, derived from  $^2S$  carbohydrates, have the absolute configuration (1'R,8R,10R) [60]. The structure for compound 99 was confirmed by X-ray analysis [48] (Scheme 17).

The reaction was extended to 1,2-dihydropyrazolo[3,4-d]-pyridazine derivatives 108, which was transformed into azomethine imines 109, 111, 113, and 115, followed by treatment with methyl acrylate to give 110, 112, 114, and 116, respectively. However, in this case 116 was formed as pure (1'S,6S,8S) stereoisomer, 112 only in 70% de, 113 in 60% de, while 110 was obtained as a racemic mixture [61] (Scheme 18).

1,3-Cycloaddition of chiral azomethine imines to heterocyclic dipolarophiles, such as substituted *p*-toluyl phthalimide, proceeds nonstereospecifically. For example, the compound **108** was transformed into azomethine imine *in situ* followed by treatment with *N*-(*p*-toluyl)-phthalimide to give *C*-nucleosides **117-120**, each of them as a mixture of two diastereoisomers [61] (Scheme 19).

In order to test the generality of this stereochemical approach, compound 108 was treated with enantiomeric sugars, followed by addition of methyl acrylate. For example, D-glyceraldehyde and L-glyceraldehyde give enantiomers 121 and 122, from D-arabinose and L-arabinose enantiomeric 123 and 126, and 124 and 125 (Scheme 20), 127 and 128, and 129 and 130 (Scheme 21) were obtained [62].

On the other hand, the cycloaddition of 108 to maleimide is non-stereospecific leading to formation of 131 and 132, each of them being a mixture of diastereoisomers (Scheme 22). This can be explained by the tautomerization process 133 = 134 = 135 = 136 in which pyrrole ring is involved. (Scheme 23). On the other hand, when N-methylmaleimide is used, the enantiomeric pairs

Scheme 17

CHO
$$H = R O H$$

$$(H = O H)_{n}$$

$$CH_{2}O H \quad n = 2,3$$

$$D_{-ribose, D-xylose, D-galactose}$$

$$D_{-glucose, D-galact$$

of sugars produce the enantiomeric pairs of *C*-nucleosides: **137** and **138**, **139** and **141**, **140** and **142** (Scheme 24). Also in these examples, the configuration is dependent upon the configuration at carbon atom next to the alde-

R = sugar chain

hydo group of sugar component. This is demonstrated by transformation of 108 with D-xylose and D-lyxose into 143 and 144 with opposite configurations on newly formed centers [62] (Scheme 25).

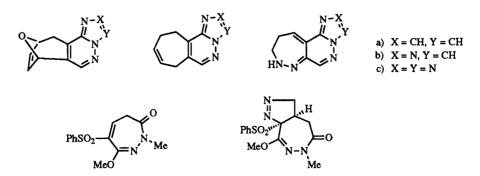
Scheme 24

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# vii) Conclusion.

In conclusion, by 1,3-dipolar cycloadditions of diazoalkanes to pyridazine derivatives several bicyclic and polycyclic pyrazolopyridazine systems were obtained (Scheme 26). These systems can be transformed either thermally or photochemically into a series of systems shown on Sheme 27, and by asymmetric 1,3-dipolar cycloaddition of chiral azomethine imines two or three new chiral centers can be introduced with high degree of regio- and diastereoselectivity into heterocyclic systems (Scheme 28).

Scheme 26



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a) X = CH, Y = CH
 b) X = N, Y = CH
 c) X = Y = N

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